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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.007 Å R factor = 0.055 wR factor = 0.138 Data-to-parameter ratio = 8.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-(Benzotriazol-1-yl)-1-p-tolylpropan-1-one

In the molecule of the title compound, $C_{16}H_{15}N_3O$, the benzotriazole system is planar. In the crystal structure, intermolecular $C-H\cdots O$ hydrogen bonds link the molecules into chains. The packing is further stabilized by $C-H\cdots \pi$ interactions.

Comment

1*H*-Benzotriazoles are an important class of compounds because of their wide use in synthetic organic chemistry and pharmaceutical sciences. They are also found to exhibit a broad spectrum of pharmacological activities such as antifungal, antitumor and antineoplastic activities (Chen *et al.*, 2005). As part of a search for new benzotriazole compounds with higher bioactivity, the title compound, (I), was synthesized and its structure is presented here.



In the molecule of the title compound, (I) (Fig. 1), the bond lengths and angles are within normal ranges (Allen *et al.*, 1987). All the bonds show a characteristic length intermediate between those of single and double bonds.

Rings A (atoms N1–N3/C10/C11), B (C10–C15) and C (C1–C6) are, of course, planar. The benzotriazole system is also planar, having a total puckering amplitude, $Q_{\rm T}$, of 0.0193 (3) Å (Cremer & Pople, 1975). The dihedral angle between rings A and B is 0.6 (2)°, while that between the benzotriazole system and ring C is 78.3 (2)°.

As can be seen from the packing diagram (Fig. 2), intermolecular C-H···O hydrogen bonds (Table 1) link the molecules into chains. The packing is further stabilized by C-H··· π interactions (Table 1).

Experimental

To a solution of Mannich base 3-(dimethylamino)-1-*p*-tolylpropan-1one (1.9 g, 0.01 mol) in water (50 ml) was added benzotriazole (1.19 g, 0.01 mol). The mixture was heated under reflux for 4 h. The resulting precipitate was purified by recrystalization from ethyl acetate–petroleum ether (1:2 ν/ν). Colorless single crystals suitable for X-ray diffraction study were obtained by slow evaporation of an

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Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.



Figure 2

A packing diagram of (I). Hydrogen bonds are shown as dashed lines.

acetone-ethanol (1:2) solution at room temperature over a period of several days.

Crystal data

 $\begin{array}{l} C_{16}H_{15}N_{3}O\\ M_{r}=265.31\\ Monoclinic, P2_{1}\\ a=11.565~(4)~\AA\\ b=5.7481~(18)~\AA\\ c=11.666~(4)~\AA\\ \beta=117.309~(5)^{\circ}\\ V=689.1~(4)~\AA^{3} \end{array}$

Data collection

Siemens SMART 1000 CCD areadetector diffractometer ω scans Absorption correction: multi-scan

(SADABS; Sheldrick, 1996) $T_{min} = 0.962, T_{max} = 0.993$ Z = 2 $D_x = 1.279 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.08 \text{ mm}^{-1}$ T = 294 (2) K Needle, colorless $0.47 \times 0.11 \times 0.08 \text{ mm}$

3769 measured reflections 1492 independent reflections 1031 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.028$ $\theta_{\text{max}} = 26.1^{\circ}$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0589P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	+ 0.1105P]
$vR(F^2) = 0.138$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} = 0.020$
492 reflections	$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
81 parameters	$\Delta \rho_{\rm min} = -0.11 \text{ e } \text{\AA}^{-3}$
I-atom parameters constrained	

Table 1	
Hydrogen-bond ge	cometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C2-H2A\cdots Cg1^{i}$	0.93	2.86	3.428	121
$C8-H8A\cdots Cg2^{ii}$	0.97	2.99	3.930	165
$C12 - H12A \cdots Cg3^{iii}$	0.93	2.86	3.534	130
$C13-H13A\cdots O1^{iii}$	0.93	2.54	3.435 (5)	161

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z$; (ii) x, y - 1, z; (iii) $-x + 1, y - \frac{1}{2}, -z - 1$. Cg1, Cg2 and Cg3 denote the centroids of the A, C and B rings, respectively

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C–H = 0.93, 0.97 and 0.96 Å for aromatic, methylene and methyl H atoms, respectively, and with $U_{\rm iso}(\rm H) = xU_{eq}(\rm C)$, where x = 1.5 for methyl and x = 1.2 for all other H atoms. In the absence of significant anomalous scattering effects, Friedel pairs were merged.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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